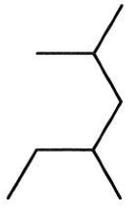
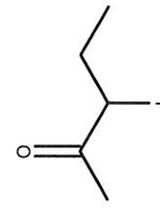
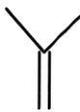
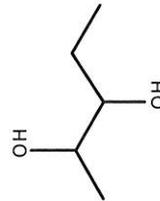
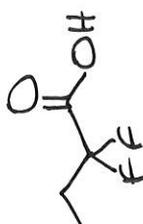


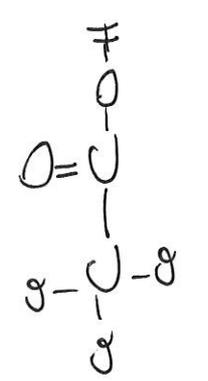
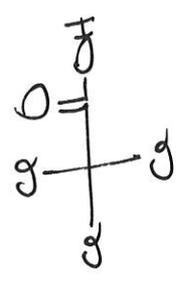
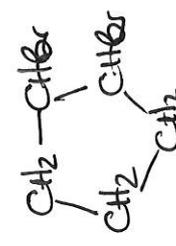
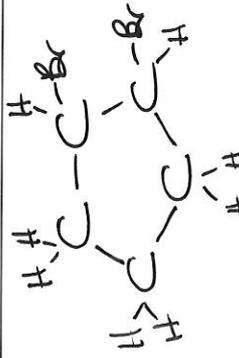
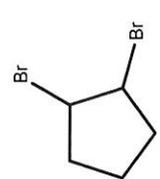
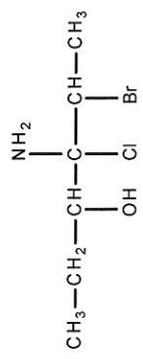
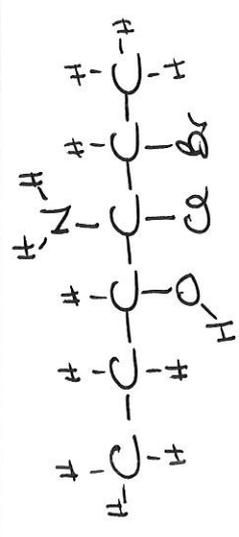
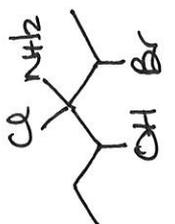
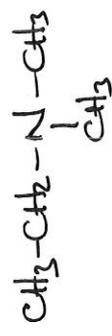
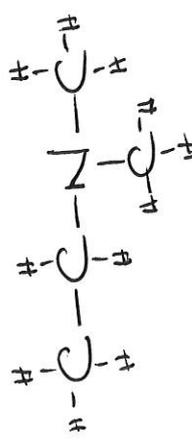
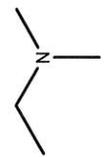


ORGANIC NOMENCLATURE 2

Q	Name	Structural formula	Displayed formula	Skeletal formula
1	2,3-dimethylbutane	$\text{CH}_3\text{CH}(\text{CH}_3)\text{CH}(\text{CH}_3)_2$		
2	4-bromo-2,2-dichloropentane			
3	butanone	$\text{CH}_3\text{-CH}_2\text{-C(=O)-CH}_3$		
4	but-2-ene	$\text{CH}_3\text{CH=CHCH}_3$		
5	butan-2-ol			

6	2,4-dimethylhexane	$ \begin{array}{c} \text{CH}_3 \\ \\ \text{CH}_3-\text{CH}-\text{CH}_2-\text{CH}-\text{CH}_2-\text{CH}_3 \\ \quad \\ \text{CH}_3 \quad \text{CH}_3 \end{array} $	
7	pentanenitrile	$\text{CH}_3-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{CN}$	
8	3-bromopropanal	$\text{CH}_2\text{BrCH}_2\text{CHO}$	
9	3-iodopentan-2-one	$ \begin{array}{c} \text{O} \\ \\ \text{CH}_3-\text{C}-\text{CH}-\text{CH}_2-\text{CH}_3 \\ \\ \text{I} \end{array} $	
10	2-methylbutylamine	$ \begin{array}{c} \text{CH}_3 \\ \\ \text{CH}_3-\text{CH}_2-\text{CH}-\text{CH}_2-\text{NH}_2 \end{array} $	

11	methylpropene	$\begin{array}{c} \text{CH}_3 \\ \\ \text{CH}_2 = \text{C} - \text{CH}_3 \end{array}$	
12	4-methylpent-2-ene	$\begin{array}{c} \text{CH}_3 \\ \\ \text{CH}_3 - \text{CH} = \text{CH} - \text{CH}_2 - \text{CH}_3 \end{array}$	
13	1,1,2-tribromoethane	$\text{CHBr}_2 - \text{CH}_2\text{Br}$	
14	pentane-2,3-diol	$\begin{array}{c} \text{OH} \\ \\ \text{CH}_3 - \text{CH} - \text{CH} - \text{CH}_2 - \text{CH}_3 \\ \\ \text{OH} \end{array}$	
15	2,2-difluorobutanoic acid	$\text{CH}_3 - \text{CH}_2 - \text{CF}_2 - \text{COOH}$	

16	trichloroethanoic acid	CCl_3COOH		
17	1,2-dibromocyclopentane			
18	3-amino-2-bromo-3-chloro-4-hydroxyhexane			
19	dimethylethylenimine (N,N-dimethylethylenimine)			
20	3-oxobutanal		