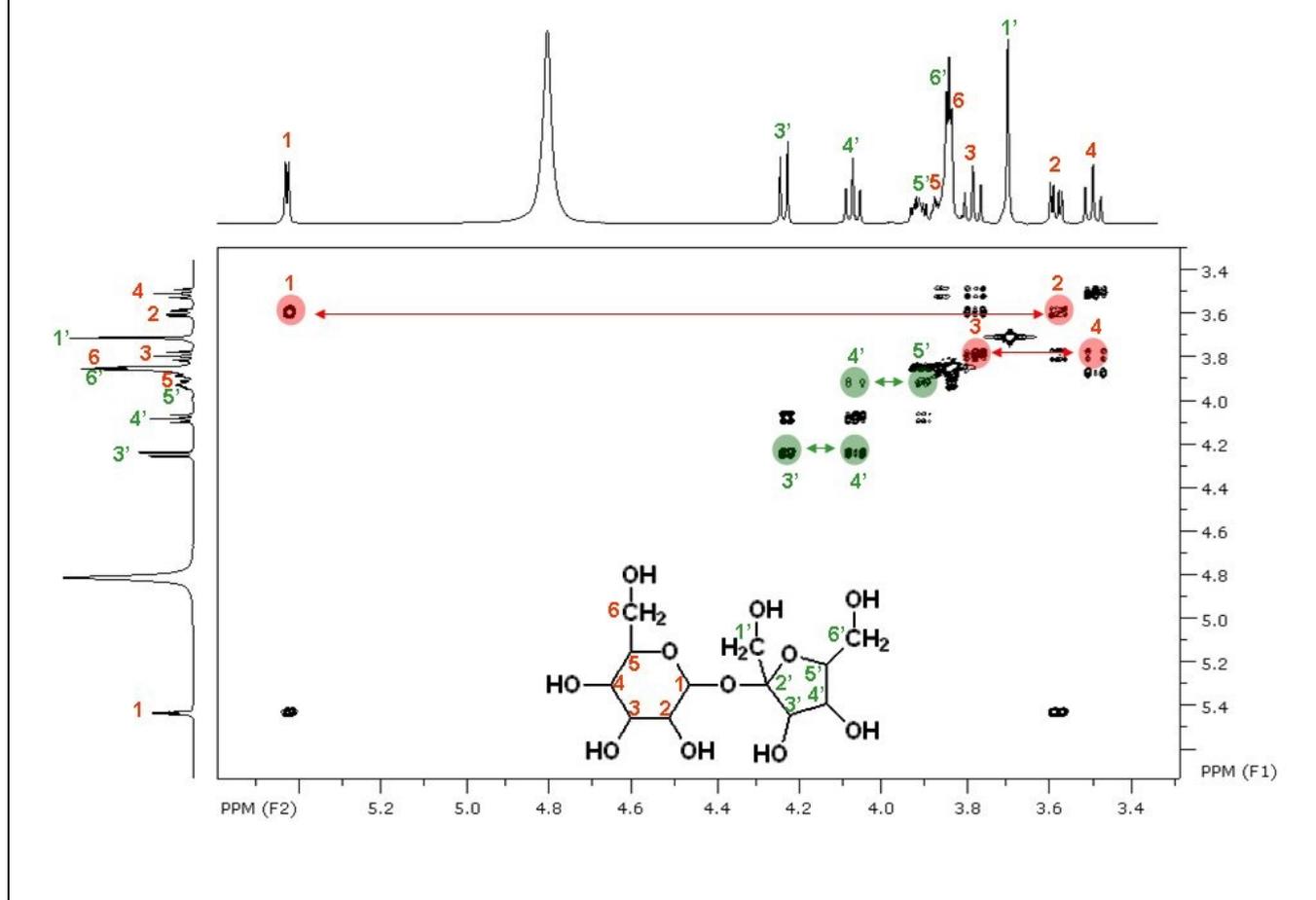


NMR

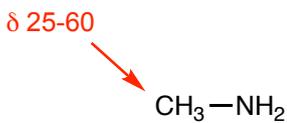
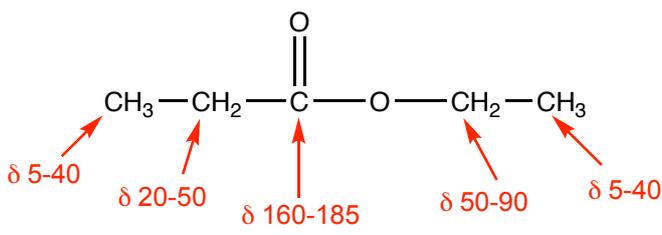
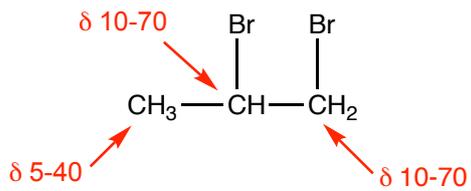
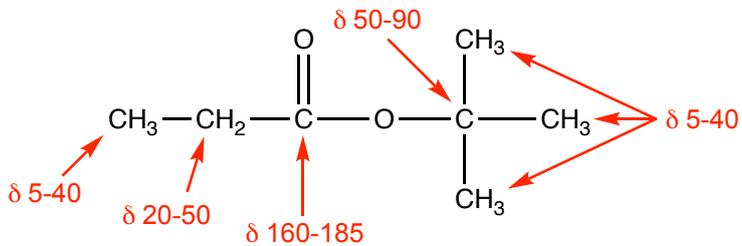
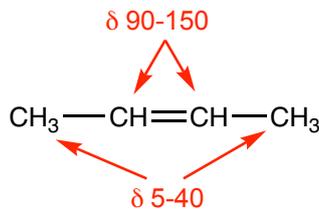
SPECTROSCOPY



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TASK 1 – Predicting ^{13}C NMR spectra

Compound	Structure and chemical shift (δ)	Number of signals
2-bromo-2-methylbutane		4
methylpropene		3
propene		3
2-chloropropane		2
propanone		2

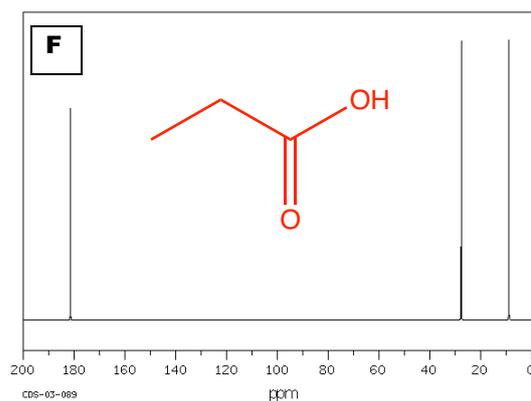
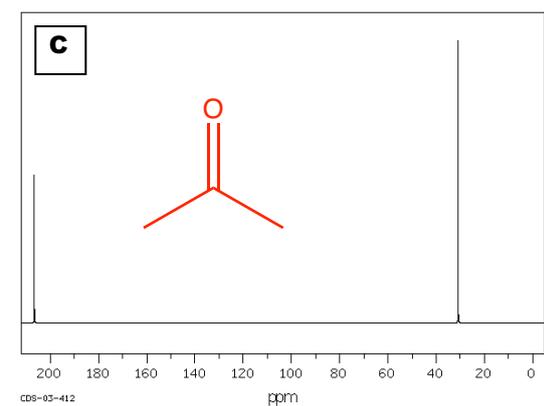
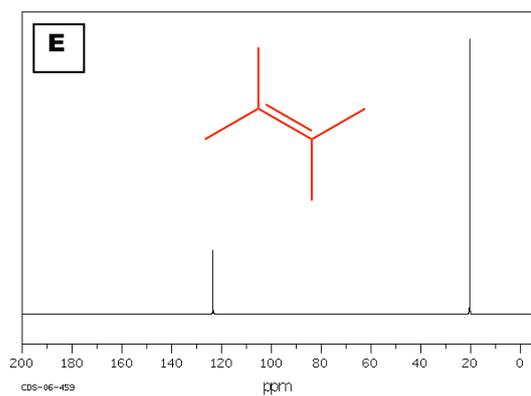
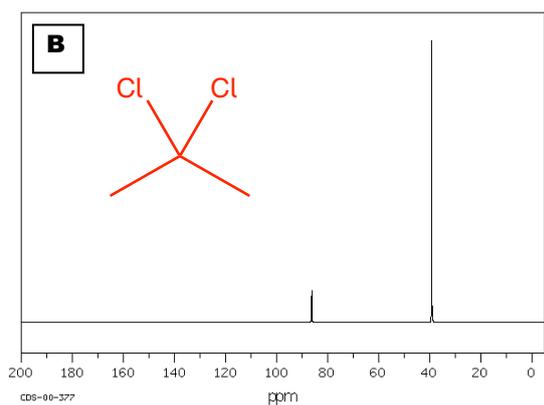
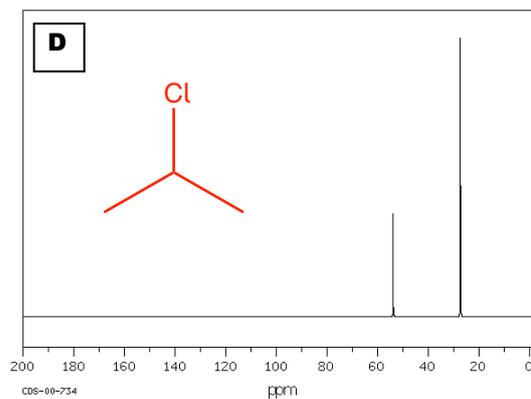
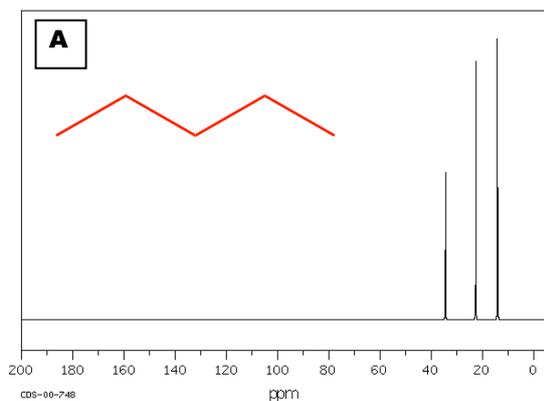
Compound	Structure and chemical shift (δ)	Number of signals
methylamine	 <p style="text-align: center;">δ 25-60 $\text{CH}_3\text{-NH}_2$</p>	1
ethyl propanoate	 <p style="text-align: center;">δ 5-40 δ 20-50 δ 160-185 δ 50-90 δ 5-40</p>	5
1,2-dibromopropane	 <p style="text-align: center;">δ 10-70 δ 5-40 δ 10-70</p>	3
dimethylethyl propanoate	 <p style="text-align: center;">δ 5-40 δ 20-50 δ 160-185 δ 50-90 δ 5-40</p>	5
but-2-ene	 <p style="text-align: center;">δ 90-150 δ 5-40</p>	2

TASK 2 – Which ^{13}C NMR spectrum is which?

For each of the following compounds:

propanoic acid pentane 2-chloropropane propanone 2,2-dichloropropane 2,3-dimethylbut-2-ene

Work out which spectrum belongs to which compound.



TASK 3a/b – Predicting ¹H NMR spectra 1

Complete the second, third and fourth columns in this table.

Compound	Structure	Number of signals	Relative intensity of signals	Splitting patterns of signals
2-bromo-2-methylbutane	$\begin{array}{c} \text{CH}_3 \\ \\ \text{CH}_3-\text{CH}_2-\text{C}-\text{CH}_3 \\ \\ \text{Br} \end{array}$	3	3 : 2 : 6	t, q, s
methylpropene	$\begin{array}{c} \text{CH}_3 \\ \\ \text{CH}_3-\text{C}=\text{CH}_2 \end{array}$	2	6 : 2 (i.e. 3 : 1)	s, s
propene	$\text{CH}_3-\text{CH}=\text{CH}_2$	3	3 : 1 : 2	d, m, d
2-chloropropane	$\begin{array}{c} \text{Cl} \\ \\ \text{CH}_3-\text{CH}-\text{CH}_3 \end{array}$	2	6 : 1	d, m
propanone	$\begin{array}{c} \text{O} \\ \\ \text{CH}_3-\text{C}-\text{CH}_3 \end{array}$	1		s
methylamine	CH_3-NH_2	2	3 : 2	s, s
ethyl propanoate	$\text{CH}_3-\text{CH}_2-\overset{\text{O}}{\parallel}{\text{C}}-\text{O}-\text{CH}_2-\text{CH}_3$	4	3 : 2 : 2 : 3	t, q, q, t
1,2-dibromopropane	$\begin{array}{c} \text{Br} \quad \text{Br} \\ \quad \\ \text{CH}_3-\text{CH}-\text{CH}_2 \end{array}$	3	3 : 1 : 2	d, m, d
dimethylethyl propanoate	$\begin{array}{c} \text{O} \quad \text{CH}_3 \\ \quad \\ \text{CH}_3-\text{CH}_2-\text{C}-\text{O}-\text{C}-\text{CH}_3 \\ \\ \text{CH}_3 \end{array}$	3	3 : 2 : 9	t, q, s
but-2-ene	$\text{CH}_3-\text{CH}=\text{CH}-\text{CH}_3$	2	6 : 2 (3 : 1)	d, q

TASK 4 – Finding the relative intensity of signals from a spectrum

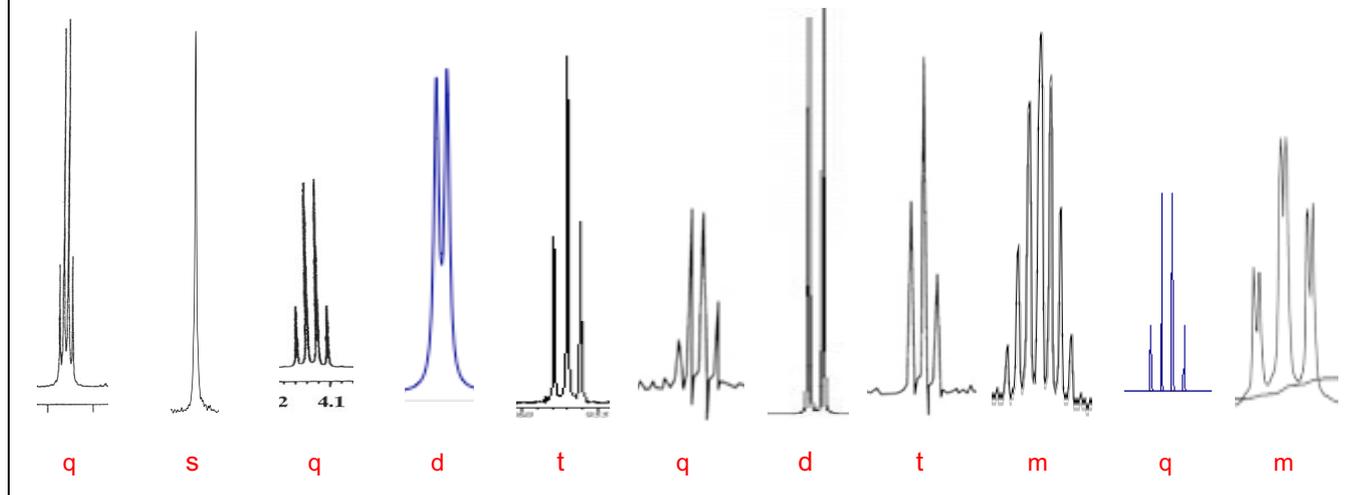
Spectrum A: 1 : 6 : 2 : 3

Spectrum B: 2 : 2 : 3 : 3

Spectrum C: 1 : 2

Spectrum D: 2 : 3

TASK 5 – Identifying splitting patterns



TASK 6 – Predicting ¹H NMR spectra

Compound	Structure, δ , relative intensity & splitting	Number of signals
2,3-dimethylbutane	<p style="text-align: center;"> δ 1.4-1.6 2 m δ 0.7-1.2 12 d </p>	2
3,4-dimethylheptane	<p style="text-align: center;"> δ 1.2-1.4 2 m δ 1.4-1.6 1 m δ 1.4-1.6 1 m δ 1.2-1.4 2 m δ 0.7-1.2 3 t δ 0.7-1.2 3 d δ 0.7-1.2 3 d δ 1.2-1.4 2 m δ 0.7-1.2 3 t </p>	9
cyclohexane	<p style="text-align: center;"> one signal δ 1.2-1.4 s </p>	1
methylcyclohexane	<p style="text-align: center;"> δ 0.7-1.2 3 d δ 1.4-1.6 1 m δ 1.2-1.4 4 m (allow q) δ 1.2-1.4 4 m δ 1.2-1.4 2 m </p>	5
2-bromo-3-chlorobutane	<p style="text-align: center;"> δ 3.1-4.2 1 m δ 3.1-4.2 1 m δ 0.7-1.2 3 d δ 0.7-1.2 3 d </p>	4

Compound	Structure, δ , relative intensity & splitting	Number of signals
pentan-3-one	<p> δ 2.1-2.6 4 q δ 0.7-1.2 6 t </p>	2
2-methylpropan-2-ol	<p> δ 0.7-1.2 9 s δ 0.5-5.0 1 s </p>	2
propanoic acid	<p> δ 2.1-2.6 δ 10.0-12.0 2 q 1 s δ 0.7-1.2 3 t </p>	3
methyl propanoate	<p> δ 2.1-2.6 δ 3.1-3.9 2 q 3 s δ 0.7-1.2 3 t </p>	3
methylpropanal	<p> δ 2.1-2.6 δ 9.0-10.0 1 m 1 s δ 0.7-1.2 6 d </p>	3

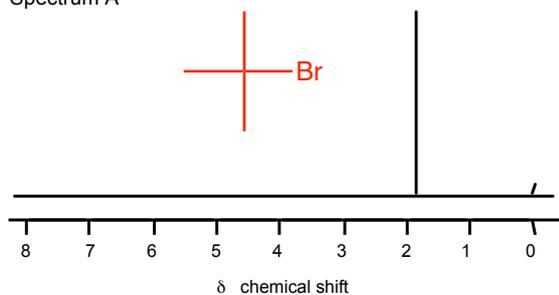
TASK 7 – Which ^1H NMR spectrum is which?

For each of the following compounds:

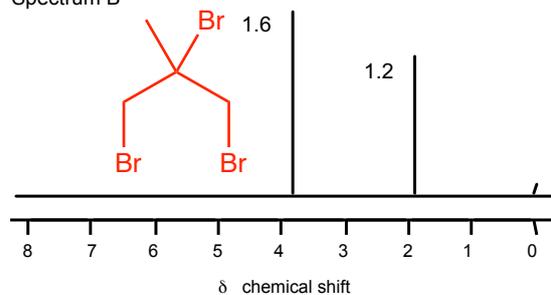


Work out which spectrum belongs to which compound.

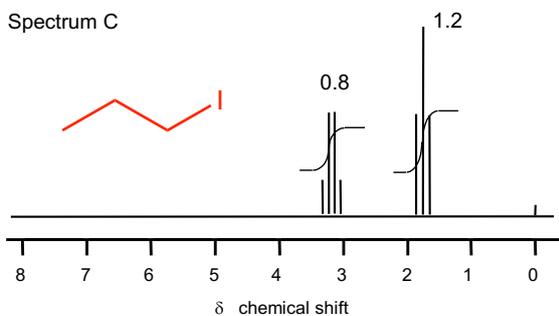
Spectrum A



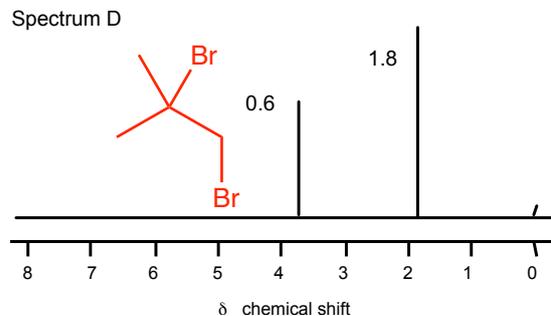
Spectrum B



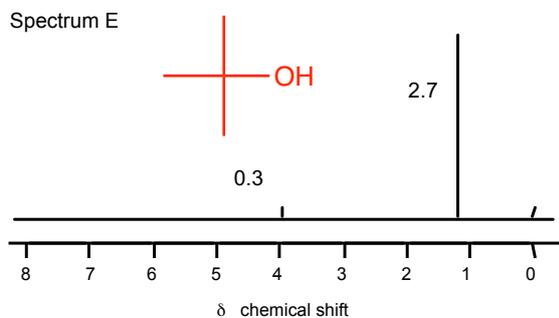
Spectrum C



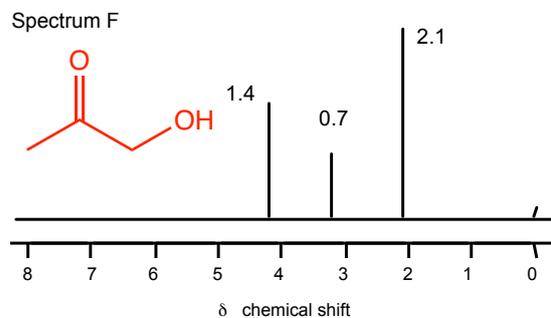
Spectrum D



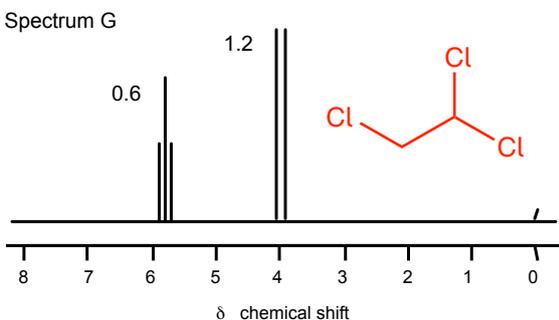
Spectrum E



Spectrum F



Spectrum G

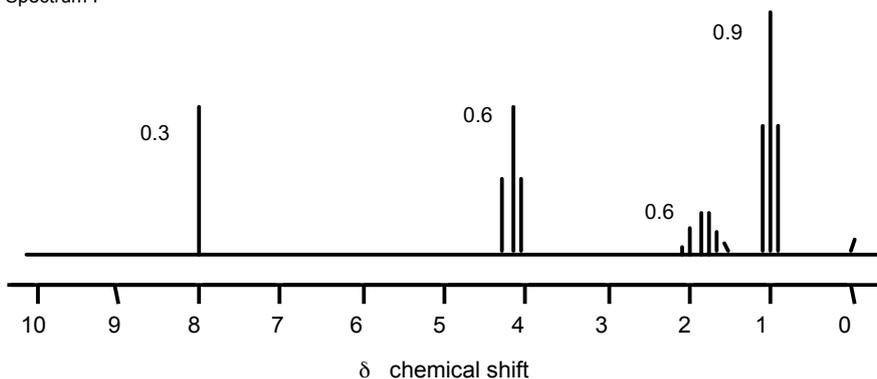


TASK 8 – Identifying compounds

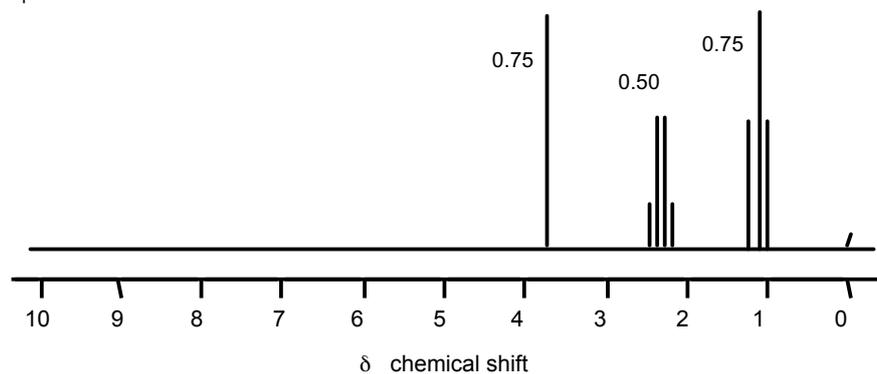
The ^1H NMR spectra of three isomers of $\text{C}_4\text{H}_8\text{O}_2$ are shown.

- Draw the structural formulae of all the isomers of $\text{C}_4\text{H}_8\text{O}_2$ that are carboxylic acids or esters.
- Indicate the number of signals, and relative intensity and multiplicity of each signal for each isomer.
- Deduce which spectrum belongs to which isomer.

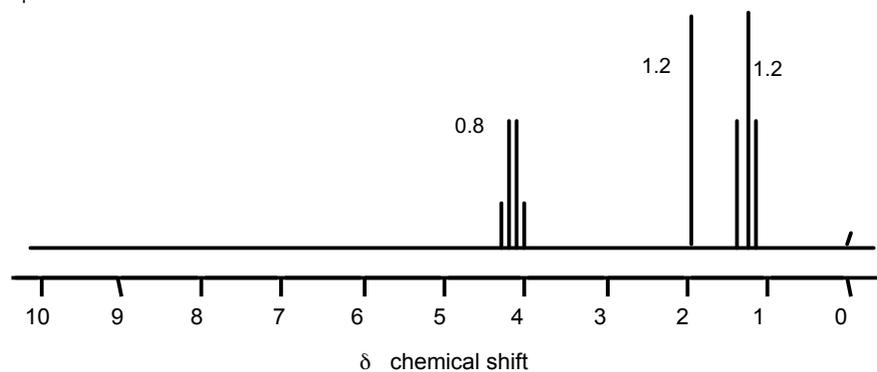
Spectrum I



Spectrum J

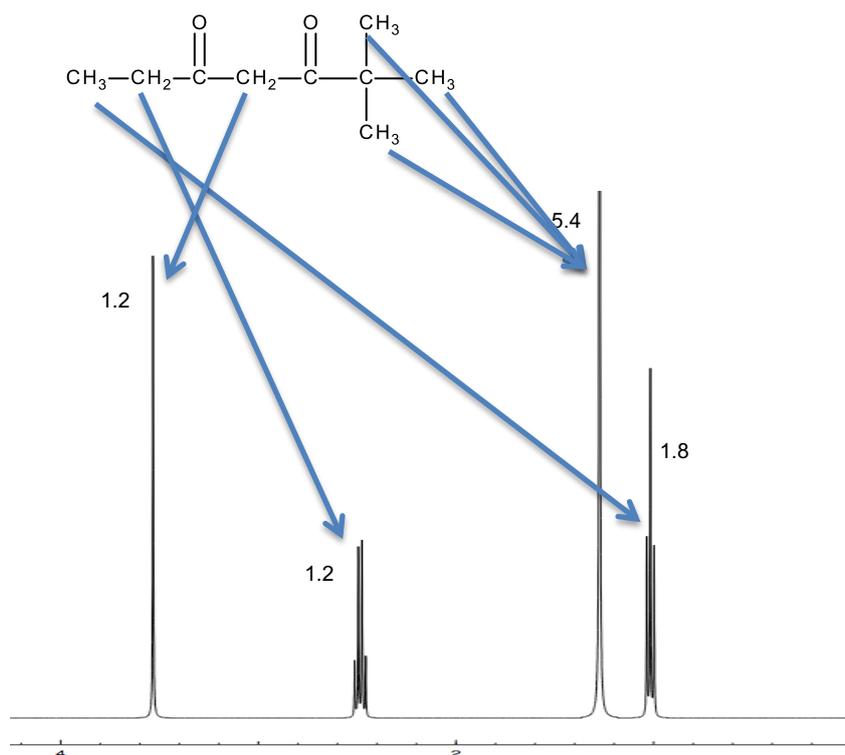


Spectrum K

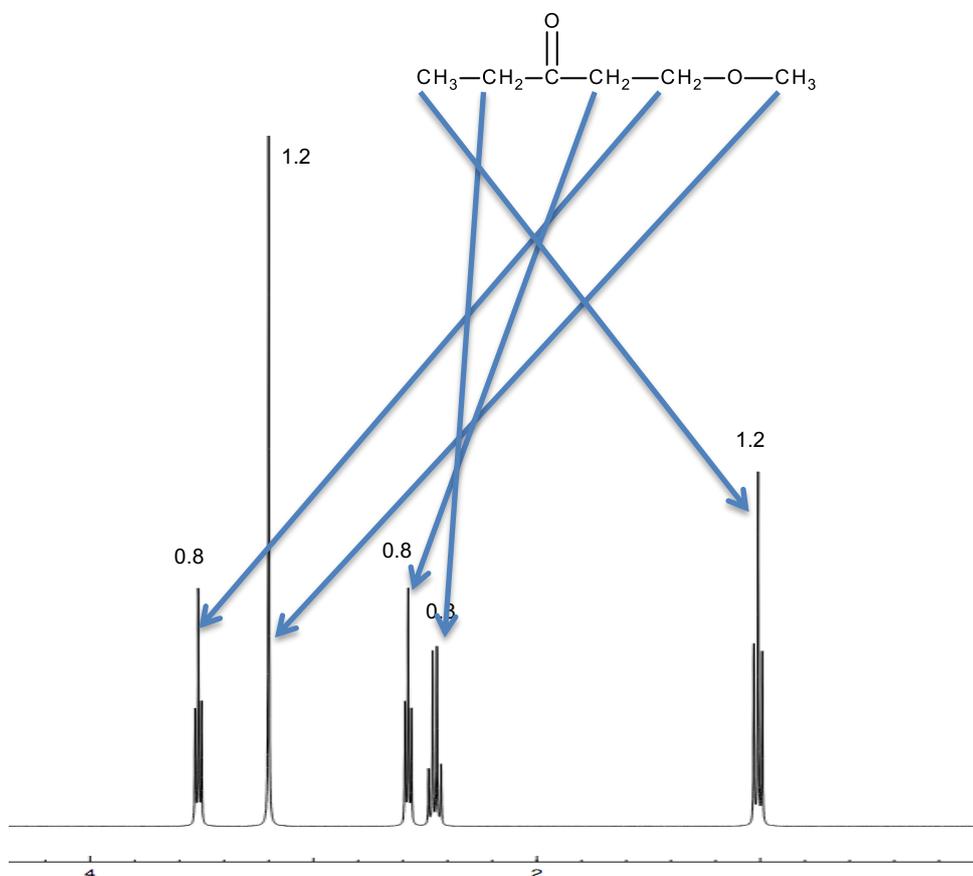


TASK 9 – Identifying compounds using ^1H NMR

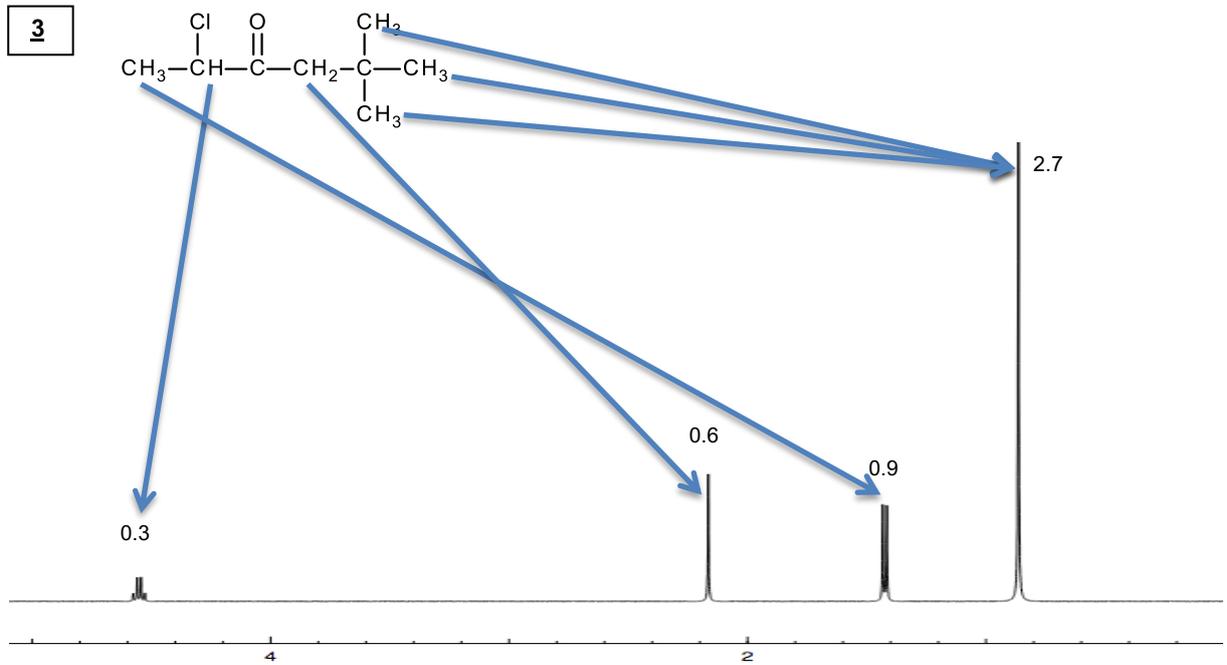
1



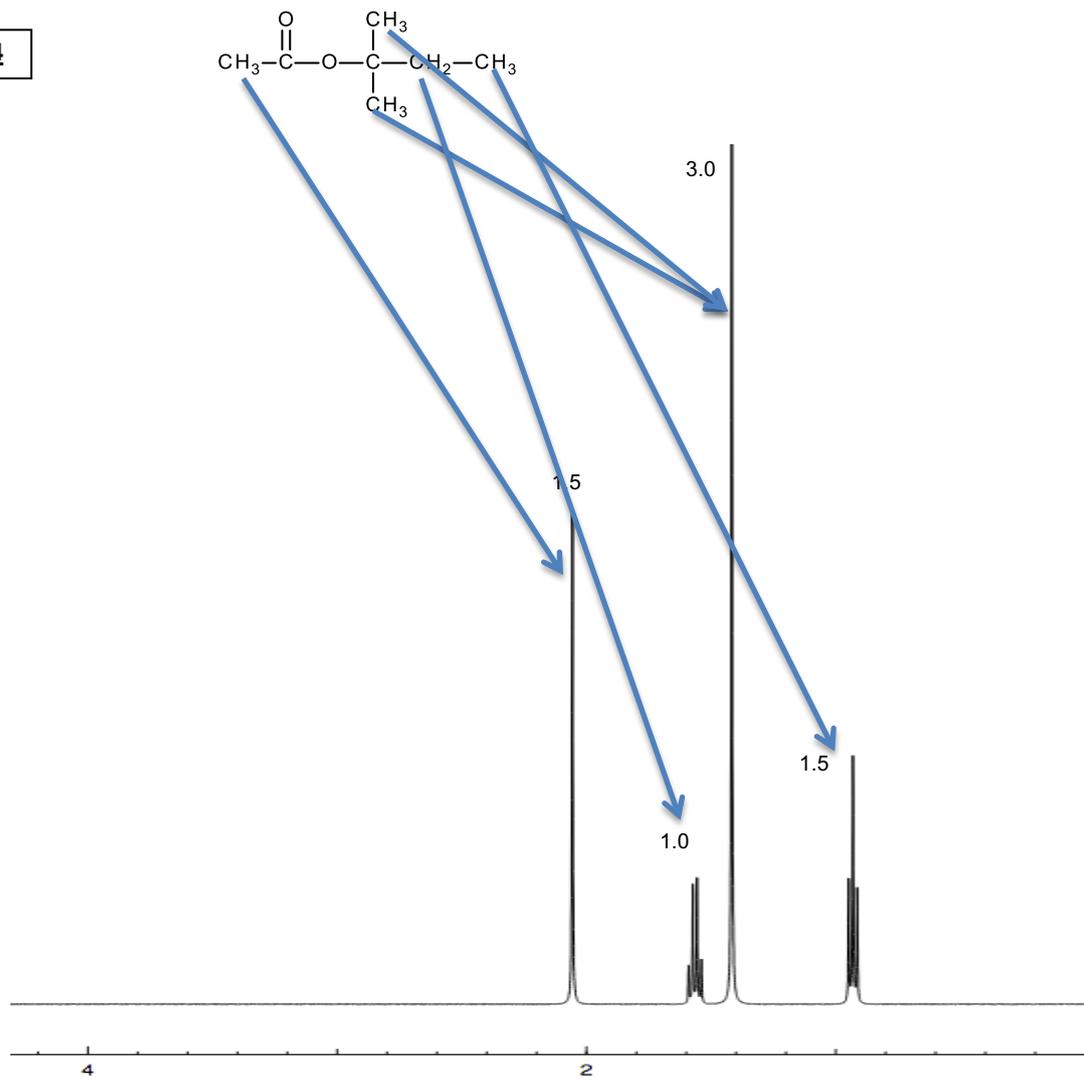
2



3

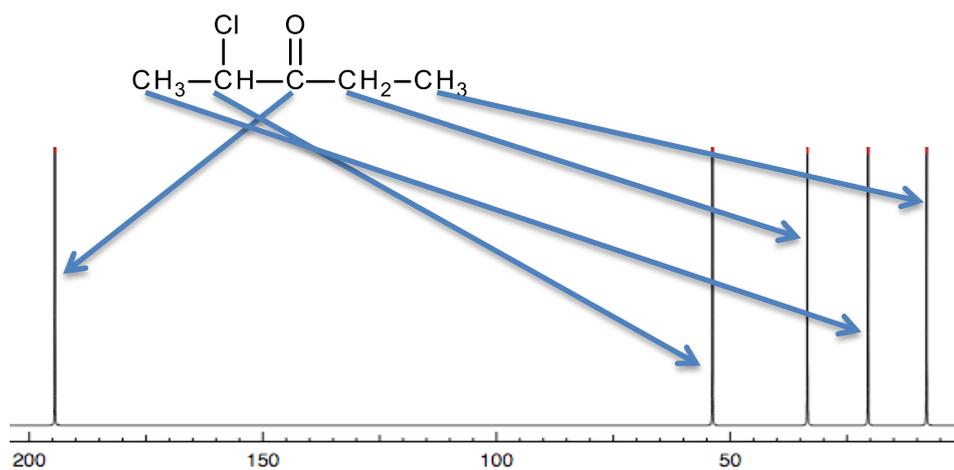
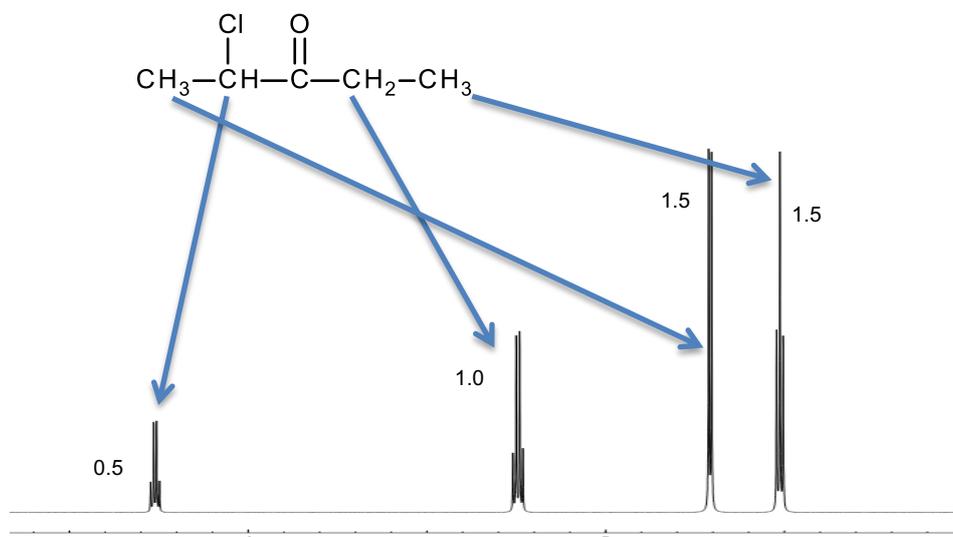


4

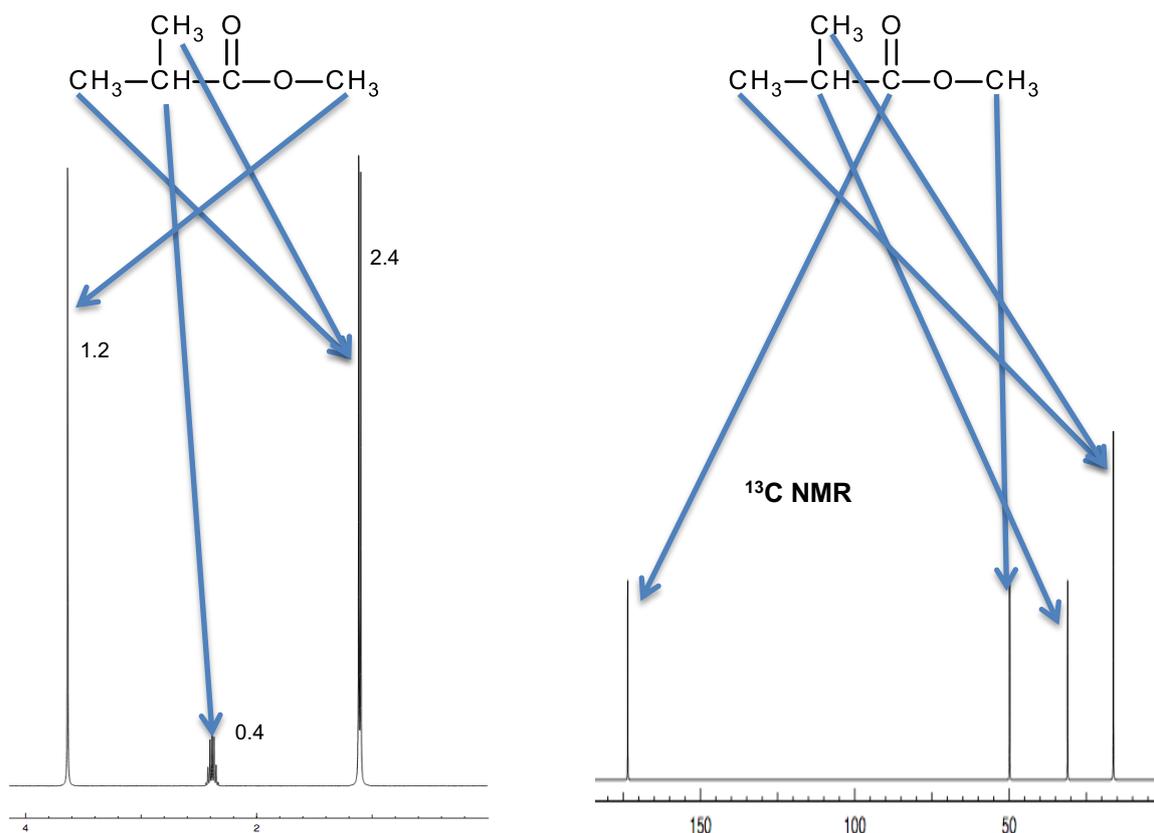


TASK 10 – Using ^1H and ^{13}C NMR together to identify compounds

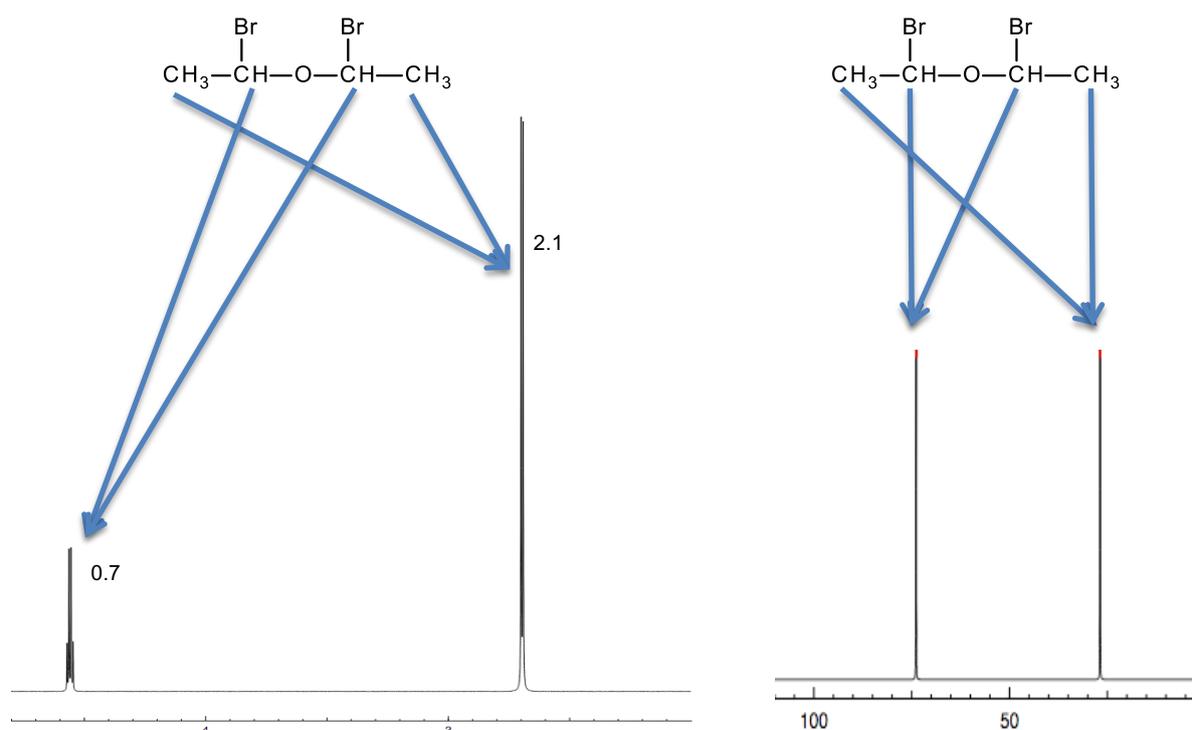
- 1 The ^1H and ^{13}C NMR spectra of $\text{C}_5\text{H}_9\text{OCl}$ are shown. Deduce the structure of the compound and then explain each signal.



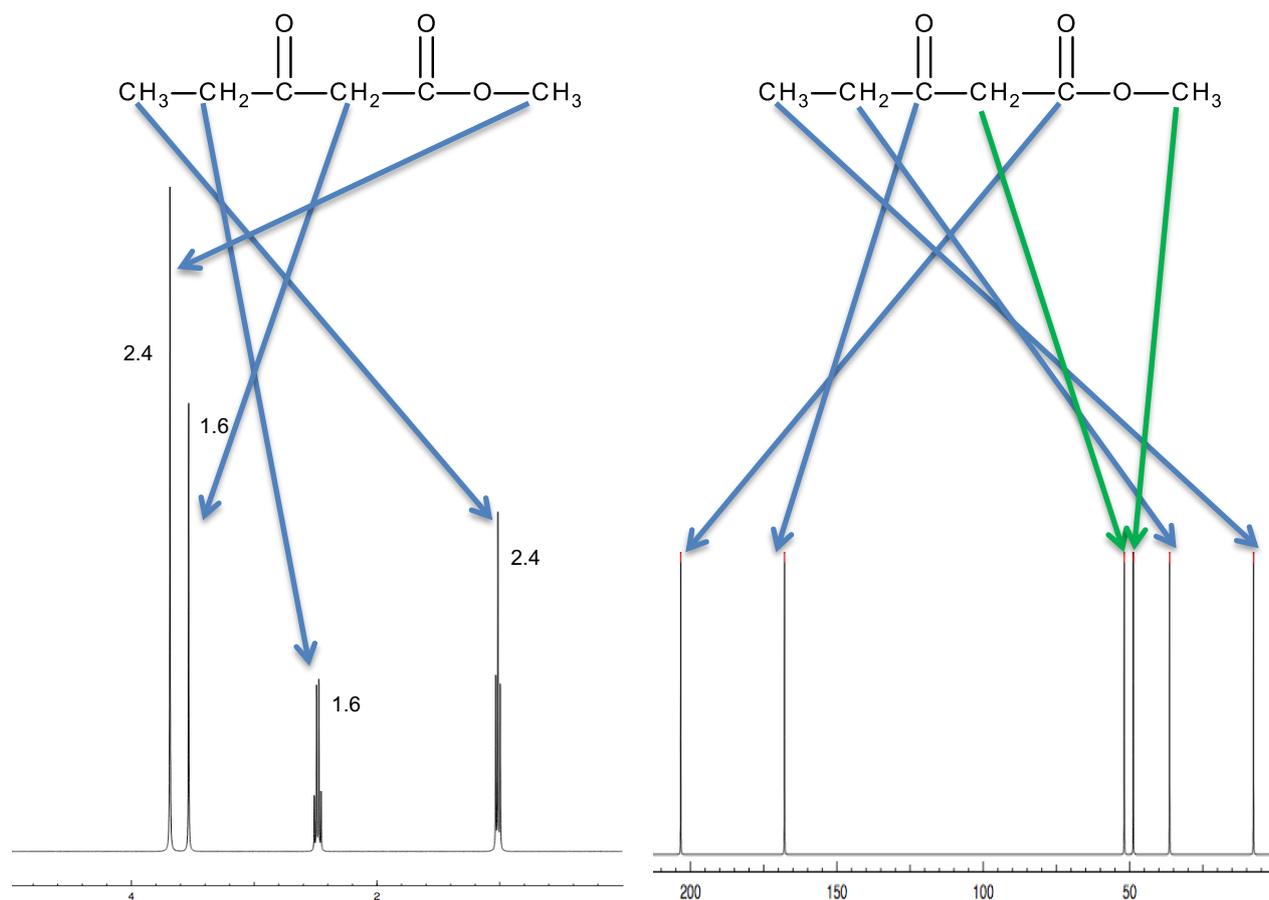
- 2 The ^1H and ^{13}C NMR spectra of $\text{C}_5\text{H}_{10}\text{O}_2$ are shown. Deduce the structure of the compound and then explain each signal.



- 3 The ^1H and ^{13}C NMR spectra of $\text{C}_4\text{H}_8\text{OBr}_2$ are shown. Deduce the structure of the compound and then explain each signal.



- 4 The ^1H and ^{13}C NMR spectra of $\text{C}_6\text{H}_{10}\text{O}_3$ are shown. Deduce the structure of the compound and then explain each signal.



lack of certainty as to which signal is which

